

Be sure to show all work, use the correct number of significant figures, circle final answers and use correct units in all problems.

1. **Circle** the color of light with the lowest energy: (1 point) *Hint: ROY G BIV*

orange blue red green yellow

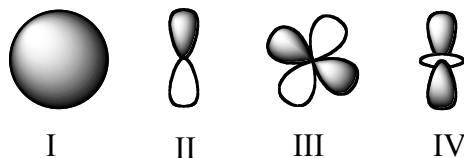
2. **Circle** the color of light with the smallest frequency: (1 point) *Hint: ROY G BIV*

orange blue red green yellow

3. Match the names on the left with their scientific discovery on the right: (1 point each)

10^3	_____	a) the number of J in a kJ
10^9	_____	b) the number of nm in a m
$1s^2 2s^2$	_____	c) paramagnetic with 2 unpaired electrons
oxygen	_____	d) electron configuration for beryllium
fluorine	_____	e) paramagnetic with 1 unpaired electron

4. Use the diagram below to answer the following questions: (2 points)



The number that corresponds to a p orbital is: _____

The number that corresponds to an orbital with no planar nodes is: _____

5. Use the information below to answer the following questions. (2 points) *Use $1s$, $2p$, $3d$, etc. for your answers.*

What type of orbital is designated $n = 2, l = 0, m_l = 0$? _____

What type of orbital is designated $n = 4, l = 1, m_l = -1$? _____

6. If the de Broglie wavelength of an electron is 112 nm, what is its velocity in m/s? The mass of an electron is 9.11×10^{-31} kg. (5 points) (*Note to physics fans: no relativity in this problem!*)

7. If an FM radio station operates at a frequency of 92.3 megahertz (MHz, or 92.3×10^6 Hz), calculate the **wavelength** of its signal in meters and the **energy** of one photon in Joules. (4 points) (+1 bonus if you can name the Portland-area radio station using this information! ☺)

Answers

1. **Circle** the color of light with the lowest energy: (1 point)

red

2. **Circle** the color of light with the smallest frequency: (1 point)

red

3. Match the names on the left with their scientific discovery on the right: (1 point each)

a

b

d

c

e

4. Use the diagram below to answer the following questions: (2 points)

II

I

5. Use the information below to answer the following questions. (2 points) *Use 1s, 2p, 3d, etc. for your answers.*

What type of orbital is designated $n = 2, l = 0, m_l = 0$? 2s

What type of orbital is designated $n = 4, l = 1, m_l = -1$? 4p

6. If the de Broglie wavelength of an electron is 112 nm, what is its velocity in m/s? The mass of an electron is 9.11×10^{-31} kg. (5 points) *(Note to physics fans: no relativity in this problem!)*

6490 m/s

7. If an FM radio station operates at a frequency of 92.3 megahertz (MHz, or 92.3×10^6 Hz), calculate the **wavelength** of its signal in meters and the **energy** of one photon in Joules. (4 points) (+1 bonus if you can name the Portland-area radio station using this information! ☺)

3.25 m

6.12×10^{-26} J